

C. U. SHAH UNIVERSITY

Winter Examination-2022

Subject Name : Inorganic Chemistry-I

Subject Code : 5SC01ICH1

Branch: M.Sc. (Chemistry)

Semester : 1

Date : 02/01/2023

Time : 11:00 To 02:00

Marks : 70

Instructions:

- (1) Use of Programmable calculator and any other electronic instrument is prohibited.
- (2) Instructions written on main answer book are strictly to be obeyed.
- (3) Draw neat diagrams and figures (if necessary) at right places.
- (4) Assume suitable data if needed.

SECTION – I

Q-1 Attempt the Following questions (07)

- Define: Magnetic susceptibility **01**
- Write formula of intensity of magnetization. **01**
- Magnetic momentum values depend on _____ and _____. **01**
- What is curie temperature? **01**
- Define : hybridisation **01**
- What are conjugated molecules ? **01**
- Molecule showing sp^2 hybridisation has bond angle of _____. **01**

Q-2 Attempt all questions (14)

- A** Evaluate the co-efficient of wave function for sp hybrid orbitals and prove that the angle between two hybrid orbital is 180° . **10**
- B** Give difference between BMO and ABMO. **04**

OR

Q-2 (14)

What is Huckel's π -electron theory? Explain its applications to Ethylene.

Q-3 Attempt all questions (14)

- A** Explain Gouy method to measure the magnetic susceptibility. **07**
- B** Write a brief note on paramagnetic substance. **07**

OR

Q-3 Attempt all questions (14)

- A** Explain diamagnetism and derive equation for diamagnetic momentum. **07**
- B** Discuss effect of temperature on different types of magnetic substance. **04**
- C** Write characteristics of diamagnetism. **03**



SECTION – II

- Q-4** **Attempt the Following questions** **(07)**
- a. Draw the structure of DMG. 01
 - b. Give uses of ammonium vanadate and its structure. 01
 - c. Write only reaction between phenol and potassium bromate. 01
 - d. Draw the structure of aluminum. 01
 - e. Who observed Mossbauer spectroscopy first and when ? 01
 - f. In Mossbauer spectroscopy which kind of ray is absorbed by absorber? 01
 - g. Define recoilless emission. 01
- Q-5** **Attempt all questions** **(14)**
- A** Draw schematic diagram of Mossbauer spectrophotometer in details **07**
 - B** Explain quadrupole splitting for Mossbauer spectroscopy. **07**
- OR**
- Q-5** **Attempt all questions** **(14)**
- A** Discuss basic principle of Mossbauer spectroscopy. **07**
 - B** Describe isomer shift with example. **07**
- Q-6** **Attempt all questions** **(14)**
- A** Discuss in detail : EDTA **08**
 - B** Write uses and physical properties of cupron. **06**
- OR**
- Q-6** **Attempt all Questions** **(14)**
- A** Write a note on oxime. **07**
 - B** Write a note on potassium bromate **07**

